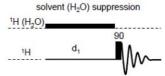
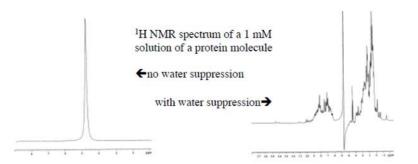


OTHER DECOUPLING: SOLVENT (H2O) SUPPRESSION

- 1H NMR in protonated solvents is problematic
- The concentration of H2O in H2O is ~55 M (1H concentration ~110 M)
- Signals from other molecules are obscured by the large H₂O signal
- One way to attenuate the H₂O signal is by "decoupling", or saturating the resonance (called solvent suppression by "saturation" or "presaturation")

 A long, selective, low power pulse is used to saturate selectively the H₂O frequency, which greatly attenuates the signal in the spectrum

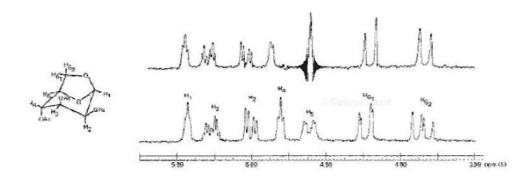




Spin Decoupling.

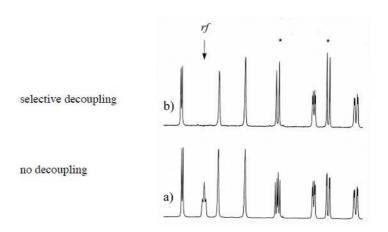
A more complex example is illustrated using the bicyclic sugar mannosan triacetate, which has a nearly first-order spectrum with numerous coupling partners.

Irradiation of H₅ @ δ 4.62 produces simplification of the resonances of its vicinal partners H₄, H_{6/1} H_{6/2} as well as its long- range zigzag partner H₃:

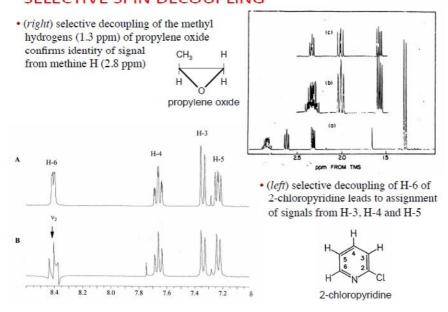


SELECTIVE SPIN DECOUPLING

- selective decoupling can identify signals from coupled nuclei, thus assisting to establish connectivity/structure
- intensities of signals from coupled nuclei are improved (due to multiplet collapse)
 integrals are unchanged for signals from coupled nuclei



SELECTIVE SPIN DECOUPLING



BROAD-BAND DECOUPLING

- Substantial spectral simplification (multiplets collapse to singlets)
- · Substantial signal-to-noise gains
- · Signal-to-noise gains arise from two sources:

